# Explore static electricity with sticky tape

Charge it



A teacher laughs as the two tapes he holds in his hands repel!

# Introduction

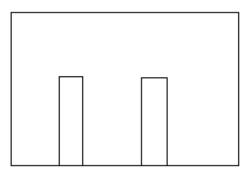
Use Scotch Magic Tape<sup>™</sup> to determine the electric charge of objects.

# Material

A roll of 3/4 inch (1.5 cm) wide Scotch Magic Tape<sup>TM</sup> Optional, scrap wood or smooth cardboard at least 20 cm (9 inches) square.

# Assembly

Pull off two 10 cm (4") lengths of tape and smooth them down onto a table or other firm surface. Call these BASE tapes. Caution, removal of the tapes may mar table surfaces, if you worry about the finish on your tables, put the BASE tapes onto smooth wood or cardboard scraps.



Put two 10 cm pieces of tape down onto a surface.

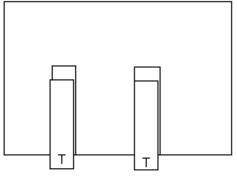
# To Do and Notice

0. Two OLD tapes.

Pull off two 10 cm lengths (4") of tape and hang them from the edge of a table to age. Let them age. Call these OLD tapes.

1. Two TOP tapes.

Pull off two 10 cm lengths of tape and stick them down on top of the two BASE tapes. Fold the end of the tapes back on themselves to make a handle. Call these TOP tapes. Optional, mark these with a T.



Put two tapes on top of the base tapes, call these top tapes, T.

Quickly pull off the two top tapes and hold one in each hand so that each hangs down vertically. (If one sticks to your hand shake it free.) Bring the two tapes near each other and observe what happens.

They repel.



Two top, T, tapes repel.

Observation 1. Two tapes treated identically repel.

The sticky side of each tape you removed was pressed against the smooth side of the BASE tape.

Here is a force that is repulsive, unlike gravity which is always attractive. Keep the tapes near each other. Over time notice that they move closer. The repulsion goes away slowly. On humid days the repulsion decreases more quickly.

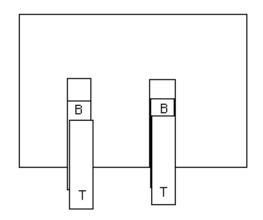
2. Two OLD tapes.

Hold one of the OLD tapes in each hand so that they hang down vertically. Bring them together. Observe what happens. Nothing happens, they neither attract nor repel. (If they repel then let them age longer.)

Observation 2. Two OLD tapes neither attract nor repel.

3. A TOP tape and a BOTTOM tape.

Pull off one 10 cm length of tape and stick it down to one of the BASE tapes, leave a handle. Call this one a BOTTOM tape and print a B on one end. Pull off another 10 cm length and stick it down on top of the BOTTOM tape. Leave a handle. Call this a TOP tape. Print a T on one end.



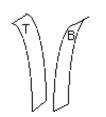
Put bottom tapes, B, on top of the base tapes, then put top, T, tapes on top of the bottom tapes.

Pull up both TOP and BOTTOM tapes. Keep them together. Stroke them gently between your fingers a few times until they no longer attract your hand.



A top, T, tape and a bottom, B tape ready to be pulled apart.

Pull the tapes apart rapidly and hold one in each hand. Bring them together. Observe what happens. They attract.



A top, T, tape and a bottom, B, tape attract.

Observation 3. Two tapes treated oppositely attract. The sticky side of one was pulled off the smooth side of the other.

4. A TOP tape and an OLD tape

Stick 10 cm of tape down on a BASE tape. Pull it up to create a TOP tape. Bring it close to an OLD tape. When the tapes are very close you may see them attract. (If there is no apparent attraction in this experiment wait until part 2 where we'll use more highly charged materials. Also remember that the charged tapes attracted your hand.)

Observation 4. Charged tapes attract uncharged tapes.

5. All combinations of TOP and BOTTOM tapes. Find a partner, pull apart two pairs of TOP and BOTTOM tapes. Test the following pairs of tapes, and notice: Observation set 5.

two TOP tapes repel two BOTTOM tapes repel Top and Bottom tapes attract

## What's Going On?

The above observations were available to scientists at the end of the 18'th century. Ben Franklin was the first scientist to devise a model to explain the observations. Today we use a model of electric charge derived from Franklin's model.

The modern model of electric charge says there are two types of electrical charge, particles may posses a positive&emdash; plus&emdash; charge or a negative &emdash; minus &emdash; charge. In normal matter these particles exist in equal numbers so that the net charge is zero. Normal matter&emdash; matter made of atoms&emdash; is uncharged.

Normal matter becomes charged when one material is rubbed against a different material. Contact between two different materials transfers electrical charge from one to the other.

The details of the charge transfer are unknown, even today. You might have read, or been told, that it is the negative charges &emdash; named electrons &emdash; which move between two objects when they are rubbed together. As of 1995 scientists do not know the details of which electric charges actually move when two tapes are pulled apart. Recent scanning tunneling microscope experiments suggest that entire atomic or molecular clusters are transferred from one material to the other. These clusters can transfer either positive or negative charge.

An editorial aside. This is a good example of how you should always ask the question "How do you know that?" Keep asking until someone can point you to an experiment that convinces you of the truth of their statement.

(An experiment that shows it is electrons that move when tapes are pulled apart would really make my job of explaining what happens during electrostatic charging easier.)

Details of the different experiments.

1. When the two TOP tapes were pulled from the BASE tapes they became charged. Since the two top tapes were treated identically they have the same charge.

Like charges repel.

We do not know whether the TOP tapes gained or lost, positive or negative charge.

The excess charge slowly decreased by combining with opposite charges in the air, or by leaking off the tape.

2. Two old tapes have equal numbers of positive and negative charge, the have no net charge and so, they neither attract nor repel. Actually there are over  $10^{23}$  positive charges in a gram of tape and an equal number of negatives.

3. When a TOP tape is Pulled from a BOTTOM tape charges move from one tape to the other. One ends up with an excess of positive charge, the other with an excess of negative.

Unlike charges attract.

In part 2 we'll find which one becomes plus and which one minus.

4. An uncharged OLD tape has many charges on it, over  $10^{23}$  positive charges and an equal number of negatives. When a charged object is brought close to the uncharged object, let's call it a plus charge for convenience, then the negative charges on the uncharged tape are attracted by the pluses and move closer while the positives are repelled and move away. The uncharged tape becomes electrically polarized. The charged tape has a stronger attraction for the nearby negatives than its repulsion for the distant positives.

Thus charged tapes attract uncharged tapes.

5. In order to explain why TOP tapes repel TOP tapes and BOTTOM tapes repel BOTTOM tapes while TOP tapes attract BOTTOM tapes requires the existence of two types of charge called plus and minus.

Ben Franklin's Model

Ben thought of electric charge as a fluid. After a time, every object, such as the OLD tapes, had a natural level of this fluid. Objects with their natural level of electrical fluid neither attracted nor repelled each other. (Observation 2.)

However, when two objects were rubbed together one object could gain an excess of this electric fluid, he called this object plus. The other would lose fluid, he called this minus. This is the origin of the terms for electrical charge we still use in our modern theory. Two objects each with an excess of fluid repelled each other.

Two objects with a deficiency of fluid repelled each other. (Observation 1, and 5)

An object with an excess attracted an object with a deficiency. (Observation 3 and 5)

Ben Franklin's model does not explain the attraction of the charged tape for the uncharged tape.(Observation 4)

Ben's choice of plus and minus as charge names has lived on after his theory. These are appropriate names because like positive and negative numbers, plus and minus charges when put together in equal amounts have a net charge which is zero.

Ben Franklin's model was eventually superseded by the modern theory of electric charge. In the modern theory particles exist with positive, negative and zero charges. For example electrons posses negative charge, protons have positive charge, and neutrons have zero net charge.

## **Continued Explorations with Tape**

Stick with me

# Introduction

When two materials were rubbed together so that each became charged, Ben Franklin did not know which one gained charge and which one lost charge. So he proposed an arbitrary standard. He said, if you rub amber (solidified tree sap) with cat fur, then, by definition, the amber becomes minus, while the fur becomes plus.

We still use his proposed standard today.

Rub a comb with a piece of wool and by Ben Franklin's definition the comb becomes negatively charged. This definition can be used to determine the charges on objects.

#### Material

A roll of Scotch Magic Tape<sup>™</sup> 3/4 inch wide (1.5 cm) a plastic or rubber comb wool, felt, or hair

Optional scrap wood or smooth cardboard at least 20 cm (9 inches) square. Optional tape holder material, you can just drape it from a table edge or hold it in a hand. A film can enough clay to fill it 1/4 full two bendable soda straws

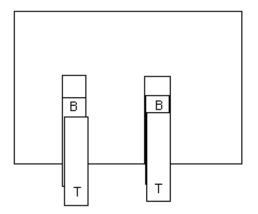
## Assembly of Optional Tape Holder

Put a lump of clay into a film can. Press the long straight end of two bendable straws into the clay. Bend over the shorter ends of each straw until they are horizontal

# To do and notice

Pull off two 10 cm (4") lengths of tape and smooth them down onto a table or other firm surface. Call these BASE tapes. Caution, removal of the tapes may mar table surfaces, if you worry about the finish on your tables, put the BASE tapes onto smooth wood or cardboard scraps.

Pull off one 10 cm length of tape and stick it down to one of the BASE tapes, leave a handle. Call this one a BOTTOM tape and print a B on one end. Pull off another 10 cm length and stick it down on top of the BOTTOM tape. Leave a handle. Call this a TOP tape. Print a T on one end.



Bottom and Top tapes are stuck down on top of the base tapes.

Pull up both TOP and BOTTOM tapes. Keep them together. Stroke them gently between your fingers a few times until they no longer attract your hand.

Pull the tapes apart rapidly and hold one in each hand.

Hang the two tapes from the horizontal arms of the straws. From our previous experiment we know that one of these is positive the other is negative.

If you are using your own comb and your hair is clean and free of oils or additives comb it, otherwise rub the comb with wool. The comb is now negative by definition.

Slowly bring the comb from far away toward the B tape.

Notice that the comb repels this tape.

Now bring it toward the T tape. The comb attracts this tape.

(Watch out! When the comb gets close to either tape it will attract the tape. So you should always observe whether the comb attracts or repels at a distance, not close-up.)

Since like charges repel, the B tape must have the same negative charge as the comb. You can now use your hanging tapes to find the charge on an object.

After the tapes have hung for a time they will lose their charge becoming OLD tapes. Bring the comb up to a 10 cm long hanging piece of old tape. Notice that the comb attracts the old uncharged tape. The old tape is not charged and yet the charged comb attracts it.

Observation 4: Charged objects attract uncharged objects.

(Well all I can really say at this point is that negatively charged objects attract uncharged objects.)

Use your tape electroscope to measure the charges on objects. Try rubbing balloons with hair, rub combs with plastic bread bags, rub acrylic with silk, rub Styrofoam with wool, and rub PVC water pipes with paper.

Remember that only repulsion can be used to reliably determine the charge on an object. Attraction can result from opposite charges or when one object is charged and the other uncharged. Objects repelled by the T tape are positive, those repelled by the B tape are negative.

The closer the comb is brought to the tape the further the tape moves. This shows that the force between the objects gets stronger as the objects get closer. When the comb approaches the tape very closely, the attraction from polarization can overwhelm the repulsion from the like charges. So watch what happens when the tapes are far from an object.

## What's Going On?

Electric charge produced by rubbing two objects together is known as triboelectricity.

The tape electroscope can be used to determine the charge on objects.

Objects charged by rubbing can be arranged in order of which becomes become positive and which become negative when they are rubbed together. This ordered list called the triboelectric series.

In general, the further apart two objects are in the triboelectric series the larger the charge they acquire when rubbed together.

There are several different versions of the triboelectric series, different versions of the series have different ordering of nearby materials, they all agree on distant materials. Here is a portion of the triboelectric series from the book "Natures Electricity.<sup>1</sup>" by Charles K. Adams, TAB books, 1987

#### Positive

Air Human flesh rabbit fur glass Human hair nylon wool cotton amber hard rubber rayon polystyrene foam (Styrofoam) polyethylene Teflon

Negative

This series predicts that a rubber comb becomes negative when rubbed with wool and positive when rubbed with a polyethylene bread bag.

Scientists do not know what charges are transferred when two materials are brought together and then separated. Images created with scanning tunneling microscopes shows that entire clusters of atoms are transferred when a tungsten electrode touches a gold surface. Contact can thus transfer positively charged clusters of atoms or negatively charged clusters. This means that rubber can become negatively charged when it is rubbed with wool either by acquiring negative ion clusters from the wool or by giving positive ion clusters to the wool. At the moment we cannot tell which is happening.

#### Why the tapes lose charge on Humid days

The two tapes on your electroscope will lose their charge slowly, over many minutes on dry days, and more rapidly, in a minute or two, on humid days. On humid days there are many so called large ions in the air as well as uncharged small water drops. A charged object will attract opposite charged ions from the air. These will slowly cancel the charge on the object. Small uncharged water drops will also be attracted. These will add a conducting coating to an object which will allow the charge to be conducted off the object.

#### etc. 1

Atoms are made of positively charged protons, negatively charged electrons and uncharged neutrons. Protons and electrons have equal and opposite charges. Neutral atoms have an equal number of protons and electrons. It is possible for atoms to gain or lose electrons and to become charged either plus or minus. Charged atoms are called ions.

#### etc. 2

One fundamental property of a particle of matter is its electric charge, another is its mass.

The mass determines the gravitational attraction. There is only one type of mass and it leads to gravitational attraction.

There are two types of electric charge. A particle may also be uncharged.

We call the two types of charge plus and minus. Combining an equal amount of plus charge to minus charge leads to an object with no net charge. Particles called electrons have one unit of negative charge.

Protons have one unit of positive charge.

Neutrons have no charge.

Photons have no charge.

## Etc 3

Actually, uncharged atoms and molecules weakly attract each other, the electrons move around so that the atoms are sometimes plus on one side and minus on the other. These polarized atoms weakly attract other similarly polarized atoms. The resulting force is called the Van der Waals force, it holds helium atoms together to form liquid helium when the temperature is cold enough, 3 K.

## Return to electrostatics

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